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Journal of the European Ceramic Society 26 (2006) 515–518

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Fabrication of structure-controlled hydroxyapatite/zirconia composite

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Available online 15 August 2005

Abstract

The structure-controlled hydroxyapatite/zirconia (HAp/ZrO2) composites were fabricated. At first, cylindrical hydroxyapatite (HAp) samples were prepared by the extrusion process, and then the extruded HAp cylindrical samples were coated with 3 mol% of Y_2O_3 partially stabilized ZrO_2 slurry, dried and aligned unidirectionally to form a composite bulk. The volume fraction of ZrO_2 in the HAp/ZrO₂ composite was estimated to be about 23 vol%. Bulk density and bending strength of the composites increased with sintering temperature. Fracture energy of $H\text{Ap/ZrO}_2$ composite sintered at 1350 °C was approximately 1.6 times higher than that of monolithic HAp. Although the bending strength of $HAp/ZrO₂$ composite prepared in this study was relatively low, it exhibited high fracture energy than HAp monolithic and a non-brittle fracture behavior was obtained without using fiber as the reinforcement.

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Keywords: Apatite; ZrO₂; Composites; Mechanical properties

1. Introduction

Hydroxyapatite (HAp) shows excellent osteoconductivity and biocompatibility with human hard tissues such as bone and teeth, and it has been used as artificial bone and dental materials.[1](#page-3-0) However, HAp shows lower mechanical strength and fracture toughness than natural bone, and hence applications of HAp to bioceramics are limited. In order to improve mechanical properties of HAp, several attempts for HAp combined with $ZrO₂$ have been made.^{2–11} Partially stabilized zirconia $(ZrO₂)$ shows biocompatibility and high mechanical properties, and it has been applied to orthopedic uses such as hip and knee joints.^{[1](#page-3-0)} Matsuno et al.^{[4](#page-3-0)} reported the fabrication of $HAp/ZrO₂$ composites by hot-pressing at 1350 °C for 15 min in Ar atmosphere under a uniaxial pressure of 30 MPa. In this processes, the spherical HAp particles were coated with 3 mol% of yttria (Y_2O_3) stabilized ZrO_2 . Bending strength and fracture toughness of $HAp/ZrO₂$ composite increased with $ZrO₂$ content, and these values of the

composite containing 50 vol% of $ZrO₂$ were 200 MPa and 3.3 MPa m^{1/2}, respectively. Shen et al.⁸ used spark plasma sintering to densify $HAp/ZrO₂$ (3 mol% of Y₂O₃ partially stabilized ZrO_2) composite at 1150 °C for 5 min in vacuum using a pressure of 50 MPa, and reported that HAp composite with 50 vol% of $ZrO₂$ had high bend strength and the fracture toughness of 440 MPa and 2.5 MPa m^{1/2}, respectively. Kim et al.¹⁰ mixed HAp powders and 40 vol% of 3 mol% of Y_2O_3 partially stabilized ZrO_2 in ethanol by ball-milling, and fabricated $HAp/ZrO₂$ composite by pressureless sintering at 1350 °C for 1 h in air using calcium fluoride (CaF₂) as sintering additives (2.5 vol%). They achieved bending strength of 180 MPa and fracture toughness of 2.3 MPa $m^{1/2}$. From the results of these studies, it is expected that mechanical properties of HAp, especially bending strength, can be improved by the reinforcement of large volume fraction of $ZrO₂$. However, it has been well known that $ZrO₂$ is a bioinert material and it does not combine with a bone since it has no osteoconductivity. Thus, the bioactivity of HAp that is one of the advantages of bioceramics would decrease with increasing the amount of $ZrO₂$. In consideration of bioactivity of the composite, it is desirable that the amount of $ZrO₂$ addition

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^{0955-2219/\$ –} see front matter © 2005 Elsevier Ltd. All rights reserved. doi:10.1016/j.jeurceramsoc.2005.07.047

to HAp composite should be as small as possible without compromising the mechanical properties. This is possible by preparing structure-controlled composites. Fibrous ceramics with a distinct fibrous texture are one of the most effective ways to improve fracture toughness of ceramics without decrease in bending strength.[12–15](#page-3-0) Hence, present authors have paid attention to the texture of fibrous ceramics, and attempted to fabricate $HAp/ZrO₂$ composite with fibrous texture.

In this study, HAp was reinforced with partially stabilized $ZrO₂$, and the structure-controlled $HAp/ZrO₂$ composites were fabricated by a two-step processing method in order to improve mechanical properties of HAp.

2. Experimental

The schematic illustration of the fabrication process of $HAp/ZrO₂$ composite in this study is shown in Fig. 1. HAp (Kishida Chemical, Japan) and 3 mol% of Y_2O_3 partially stabilized $ZrO₂$ (TZ-3Y, Tosoh, Japan) powders were used as the starting materials. Aqueous slurry of HAp powder was prepared and used for extrusion. Agar dissolved in hot water was used as the binder. Cylindrical HAp rods with a diameter of about 1.0 mm were prepared by extrusion method. $ZrO₂$

Fig. 1. Schematic illustration of fabrication process of HAp/ZrO₂ composite in this study.

Fig. 2. Optical micrograph of $HAp/ZrO₂$ compact before sintering.

powder was mixed with 3 mass% polyvinyl alcohol aqueous solution and the extruded cylindrical HAp rods were dipped into $ZrO₂$ slurry. After drying, the cylindrical HAp rods with $ZrO₂$ coating were unidirectionally aligned to form a bulk, and HAp powder was filled in between each cylindrical HAp rod in order to increase the density of $HAp/ZrO₂$ composite. The compact was formed under a uniaxial pressure of 25 MPa, followed by cold isostatic press under a pressure of 190 MPa. The optical micrograph of the compact is shown in Fig. 2. The microstructure of the compact was very similar to the microstructure of SiC/SiC composite fabricated by Ube Industries (Tyrannohex).^{[16](#page-3-0)} The compact was pressureless sintered between 1250 ◦C and 1350 ◦C in air for 12 h.

The composites were cut into rectangular bars (width: 3.5 mm, thickness: 2.5 mm, length: 25 mm) and were used for further characterizations. Bulk density of the composite was measured by Archimedes' method using ethanol. Three-point bending strength measurement (span length: 20 mm) was performed at room temperature in air with a crosshead speed of 0.5 mm/min using testing machine (AG-I, Shimadzu, Japan). The volume fraction of $ZrO₂$ and the crystalline phase of the composite were determined by X-ray diffractometry (XRD; RINT-1500, Rigaku, Japan). Element analysis of the composite was performed by electron probe X-ray micro analyzer (EPMA; JXA-800, JEOL, Japan). Microstructure and fracture surface of the composite were observed by scanning electron microscope (SEM; JSM-6400, JEOL, Japan).

3. Results and discussion

[Fig. 3](#page-2-0) shows microstructure of $HAp/ZrO₂$ composite sintered at 1350 °C for 12 h. From the result of EPMA analysis, it was found that cylindrical HAp phase was surrounded by $ZrO₂$, and HAp homogeneously existed around $ZrO₂$ layer-coated cylindrical HAp rod. The volume fraction of $ZrO₂$ in the composite determined by XRD analysis was about 23 vol%. XRD analysis indicated that the composite mainly consisted of crystalline phases such as HAp and Y_2O_3 partially stabilized $ZrO₂$. In addition, small amount of α -

Fig. 3. Microstructure of HAp/ZrO₂ composite sintered at 1350 °C.

tricalcium phosphate formed by the decomposition of HAp during sintering could be detected.

Bulk density and bending strength of $HAp/ZrO₂$ composites were shown in Fig. 4 as a function of sintering temperature. For comparison, bulk density and bending strength of monolithic HAp was also measured. Bulk density of monolithic HAp increased with sintering temperature, whereas its bending strength decreased from 120 MPa to 90 MPa with increasing sintering temperature due to grain growth at higher sintering temperatures. On the other hand, bulk density and bending strength of the composites increased with increasing sintering temperature, and the highest bending strength exhibited by composite, i.e. 100 MPa, is nearly the same as that of monolithic HAp sintered at $1300\,^{\circ}\text{C}$ and $1350\,^{\circ}\text{C}$. This is attributed to the amount of porosity of the composite (12–20%) at higher temperature. This might be attributed to the different sintering rate of H Ap and $ZrO₂$.

Fig. 5 shows the typical load–displacement curves of HAp/ZrO₂ composite sintered at different temperatures and

Fig. 4. Bulk density and bending strength of monolithic HAp and $HAp/ZrO₂$ composites sintered at various temperatures (square mark: monolithic HAp; circle mark: HAp/ZrO₂ composite; open mark: bulk density; solid mark: bending strength).

Fig. 5. Typical load–displacement curves of $HAp/ZrO₂$ composite sintered at various temperatures. Load–displacement curve of monolithic HAp sintered at $1350\,^{\circ}\text{C}$ is also shown in this figure.

monolithic HAp sintered at 1350 ◦C. It is observed that monolithic HAp fractured in brittle behavior, whereas the composite prepared in this study showed a non-brittle fracture behavior as can be seen in fiber-reinforced ceramic matrix composite. This might be due to crack deflection and bridging between cylindrical HAp phase and $ZrO₂$ coated layer, which resulted in a non-brittle fracture behavior as shown in Fig. 5. In the case of $HAp/ZrO₂$ composite sintered at 1250 °C and 1300 °C, the load–displacement curves showed small tails after first matrix cracking and large drop of load. In the case of $HAp/ZrO₂$ composite sintered at 1350 °C, the load gradually decreased with increasing crosshead displacement after reaching maximum load. The difference in load–displacement curves might be attributed to the difference in bulk density, i.e. pores and gaps in $HAp/ZrO₂$ composite.

The fracture energy of $HAp/ZrO₂$ composite and monolithic HAp sintered at 1350 ◦C was calculated from the area of load–displacement curve. The fracture energy of HAp/ZrO₂ composite sintered at 1350 °C was calculated as 140 J/m^2 , which is approximately 1.6 times higher than that of monolithic HAp.

The fracture toughness of $HAp/ZrO₂$ composite and monolithic HAp sintered at 1350 ◦C was approximately estimated from their elastic modulus and fracture energy, and fracture toughness of $HAp/ZrO₂$ composite was slightly improved in comparison with that of monolithic HAp (about 1.1 times). It could be found that the fracture energy must be increased significantly in order to increase fracture toughness of HAp/ZrO₂ composite.

4. Conclusions

In this study, HAp was reinforced with 3 mol % of Y_2O_3 partially stabilized $ZrO₂$ and structure-controlled $HAp/ZrO₂$ composites were fabricated in order to improve mechanical properties of HAp.

The volume fraction of $ZrO₂$ in the composite determined by XRD analysis was about 23 vol%. Bulk density and bending strength of the composites increased with increasing sintering temperature, and the values ranged from 3.4 g/cm^3 to 3.8 g/cm^3 and 70 MPa to 100 MPa , respectively. Bending strength of the composite sintered at 1300 °C and 1350 °C was nearly the same as that of monolithic HAp sintered at 1300 °C and 1350 °C.

Monolithic HAp fractured in brittle behavior, whereas the composite obtained in this study showed non-brittle fracture behavior as seen in fiber-reinforced ceramic matrix composite. It was considered that crack deflection and bridging between cylindrical HAp and $ZrO₂$ layer would occur in fracture, resulting in non-brittle fracture behavior. Fracture energy of $HAp/ZrO₂$ composite sintered at 1350 °C was approximately 1.6 times higher than that of monolithic HAp, and the value of $HAp/ZrO₂$ composite was about 140 J/m².

In this study, $HAp/ZrO₂$ composite with non-brittle fracture behavior and an increase in fracture energy could be obtained without using fiber as the reinforcement. In order to achieve a marked increase in bending strength and fracture energy without increasing the amount of $ZrO₂$, optimum fabrication process and sintering conditions must be considered.

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